Controlling the formation of 1 out of 64 stereoisomers using organocatalysis†

Søren Bertelsen, Rasmus L. Johansen and Karl Anker Jørgensen*

Received (in Cambridge, UK) 15th April 2008, Accepted 12th May 2008 First published as an Advance Article on the web 9th June 2008

DOI: 10.1039/b806418k

The formation of 1 out of 64 stereoisomers, *i.e.* controlling the creation of 6 stereocenters, of important optically active bicyclo[3.3.1]non-2-ene compounds, has been achieved using organocatalysis; the reaction proceeds with excellent stereocontrol and can be carried out with the generation of enantiopure products using chromatography-free procedures.

Recently, organocatalysis has proven to be a powerful strategy for the stereoselective synthesis of highly valuable building blocks, being complementary to the established methods using metals or enzymes.¹ The organocatalysts are readily available small organic molecules, which are easily synthesized or purchased and are tolerant of a great variety of benign reaction conditions.¹

In the field of synthesis of complex molecules with one or multiple stereocenters, the asymmetric catalytic strategy, using e.g. organocatalysis, has shown promising results towards the simultaneous synthesis of several stereocenters in easy multicomponent, domino or cascade procedures,² a task that was traditionally considered elusive. Examples are now available for the formation of six-membered ring systems having several stereocenters as in e.g. the substituted cyclohexenes developed by Enders et al. using a three-component domino reaction,^{2d} or Hayashi et al.'s Michael-Henry reaction sequence for the formation of optically active cyclohexane derivatives.² Recently, we have shown the development of a one-pot asymmetric reaction with the formation of five stereocenters by an intermolecular two-component reaction of α,β-unsaturated aldehydes with a dinitroalkane, leading in this particular case to the formation of highly substituted optically active cyclohexanols.3,9

In the present work we will show the development of a new organocatalytic asymmetric cascade reaction by demonstrating that two simple molecules, an α,β -unsaturated aldehyde and a tricarbonyl compound, react in a one-pot reaction to selectively form 4 new carbon–carbon bonds, providing 6 new stereocenters, and thereby 1 out of 64 possible (2^6) stereoisomers with excellent diastereo- and enantioselectivity (Scheme 1). The optically active products formed, bicyclo[3.3.1]non-2-enes, have a number of important applications and properties. The carbon skeleton in these compounds is a

Danish National Research Foundation: Center for Catalysis, Department of Chemistry, Aarhus University, DK-8000 Aarhus C, Denmark. E-mail: kaj@chem.au.dk; Fax: +45 8619 6199; Tel: +45 8942 3910



Scheme 1 Organocatalytic asymmetric two-component reaction leading to bicyclo[3.3.1]non-2-enes.

recurrent target in total synthesis of highly important biomolecules, while the racemic analogues have shown antitumor activity.

The proposed mechanism for the formation of the 6 stereocenters in the optically active bicyclo[3.3.1]non-2-ene products 4 is outlined in Scheme 2. The chiral organocatalyst 3 acts by first activating the α,β -unsaturated aldehyde 1 via an iminium ion intermediate A shown to the left in the scheme. This allows for a nucleophilic attack by one of the methylene carbon atoms of the tricarbonyl compound 2 at the β-carbon atom in A, generating the first 2 stereocenters. A cyclization initiated by the second activated methylene of the tricarbonyl compound leads to intermediate **B**. It is likely, yet unclear, that the cyclization is preceded by hydrolysis of the catalyst. Intermediate **B** can undergo a stereoselective conjugate addition by another molecule of tricarbonyl compound 2 leading to the formation of C. The diastereocontrol of this second addition is probably governed by the stereocenter bearing R in the second catalytic cycle to the right in Scheme 2. A final ring closure reaction, between the last free activated methylene and the central ketone leads to the highly functionalized product D (with 7 stereocenters), which is in tautomeric equilibrium with the more stable enol form 4, by reason of strong intramolecular hydrogen bonding. The piperidine might act as a catalyst for several of the reaction steps in the second part of the reaction course.

A series of experiments were performed to develop the optimal reaction conditions for the formation of the optically active bicyclo[3.3.1]non-2-ene products. The optimization investigations gave the following standard reaction conditions: dimethyl 3-oxapentanedioate **2** (0.50 mmol) was added at room temperature to a stirred solution of (S)-(-)- α , α -diphenyl-2-pyrrolidinemethanol trimethylsilyl ether **3** (0.025 mmol), benzoic acid (0.025 mmol) and the α , β -unsaturated aldehyde **1** (0.25 mmol) in toluene (125 μ L). After complete consumption of the α , β -unsaturated aldehyde, as monitored by ¹H NMR spectroscopy, MeOH (1.0 mL) and piperidine (0.05 mmol) were added and the reaction was stirred at 40 °C until full consumption of **2** (as indicated by TLC). Table 1

[†] Electronic supplementary information (ESI) available: Full experimental details. CCDC 682910 and 687505. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/b806418k

Scheme 2 Proposed organocatalytic cycles for the formation of 1 out of 64 stereoisomers by one-pot organocatalytic cascade reaction of an α, β -unsaturated aldehyde with a tricarbonyl compound.

shows the results obtained for the reaction of a number of different α,β -unsaturated aldehydes 1 with dimethyl 3-oxapentanedioate 2 in the presence of 3 (10 mol%) and piperidine (20 mol%) as the catalysts (Scheme 3).

Scheme 3 Reaction of different α, β -unsaturated aldehydes 1 with dimethyl 3-oxapentanedioate 2 in the presence of 3 (10 mol%) and piperidine (20 mol%) as the catalysts.

As it appears from Table 1, various α , β -unsaturated aldehydes have successfully been subjected to the reaction conditions giving a broad range of optically active bicyclo[3.3.1]non-2-enes **4**. Aliphatics (linear and branched, including heteroatoms and olefins), aromatics, heteroaromatics and esters are all well tolerated as substituents in the

reaction. For the simple aliphatic α , β -unsaturated aldehydes (entries 1–3), the optically active products 4a-4c are formed in good yields, after the multiple reaction steps, and for the two former products, only one diastereoisomer could be detected and both with more than 94% ee. For the optically active bicyclo[3.3.1]non-2-enes having an ester or olefin as R, 4d and 4e, respectively, excellent results are also obtained (entries 4, 5). The product diversity can be expanded to also include aromatic and heteroaromatic substituents as shown for 4f-4i. The aromatic compounds, both unsubstituted, and *para*- and *ortho*-substituted phenyls (entries 6-9), gave excellent yields with up to 93% isolated yield, one diastereoisomer and up to 96% ee. The presence of a heteroaromatic substituent, such as furyl, 4h, led to an 86% yield, 94:6 dr and 90% ee.

Table 1 Organocatalytic asymmetric cascade reaction of different α,β -unsaturated aldehydes with dimethyl 3-oxapentanedioate

Entry	R		Yield (%)	Dr^a	Ee ^b (%)
1	Et	4a	48	>99:1	94
2	<i>i</i> -Pr	4b	65	>99:1	96
3	$n-C_7H_{15}$	4c	69	88:12	95
4	EtO ₂ C	4d	38	>99:1	89
5	(Z)-3-Hexenyl	4e	51	94:6	94
6	Ph	4f	70	>99:1	93
7	p-MeOPh	4g	93	92:8	91
8	2-Furyl	4h	86	94:6	90
9	o-BrPh	4i	86	>99:1	96

^a Dr measured by ¹H NMR spectroscopy. ^b Ee measured by HPLC using a Daicel Chiralpak AD column.

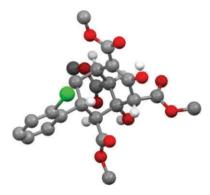


Fig. 1 X-Ray structure of (1R,4R,5S,6S,7R,9S)-tetramethyl 7-(2bromophenyl)-3,5-dihydroxybicyclo[3.3.1]non-2-ene-2,4,6,9-tetracarboxylate 4i. Colour coding: white: hydrogen; grey: carbon; red: oxygen; green: bromine. Some hydrogen atoms are omitted for clarity.

This new organocatalytic reaction can easily be scaled up to gram scale and isolation of the optically active bicyclo[3.3.1]non-2-enes 4 was performed using chromatography-free procedures by crystallizing the product after finishing the reaction. Using this procedure, product 4a was isolated in 3 g quantity (44% yield, compared to 48% on the mg scale) and as an enantiopure compound; i.e. dr > 99 : 1 and > 99% ee. The other products could also be isolated as enantiopure compounds after recrystallization.

The optically active bicyclo[3.3.1]non-2-enes having 6 stereocenters are set up for the introduction of 2 additional stereocenters in the ring systems and e.g. initial hydrogenation studies of the enol in 4a shows promising results and is currently under investigation.

The absolute configuration of tetramethyl 7-(2-bromophenyl)-3,5-dihydroxybicyclo[3.3.1]non-2-ene-2,4,6,9-tetracarboxylate 4i, was unambiguously established as (1R,4R,5S,6S,7R,9S) by X-ray crystallographic analysis (Fig. 1).8

Organocatalysis has been taken to a new level, allowing the selective formation of 4 new carbon–carbon bonds, providing 6 new stereocenters, one of which is quaternary, leading to the controlled synthesis of 1 out of 64 possible stereoisomers by mixing two simple molecules, an α,β-unsaturated aldehyde and a tricarbonyl compound. The products generated, optically active bicyclo[3.3.1]non-2-enes, are formed in high yield, and with excellent diastereo- and enantiocontrol and performing the reaction on the gram scale leads to optically pure products. The optically active bicyclo[3.3.1]non-2-enes are bicyclic carbon skeletons which are precursors for important biomolecules with antitumor activity.

This work was made possible by a grant from The Danish National Research Foundation and OChemSchool. Thanks are expressed to Dr Jacob Overgaard for the X-ray analysis.

Notes and references

- 1. For recent reviews on organocatalysis see e.g.: (a) P. L. Dalko and L. Moisan, Angew. Chem., Int. Ed., 2004, 43, 5138; (b) A. Berkessel and H. Gröger, Asymmetric Organocatalysis, VCH, Weinheim, Germany, 2004; (c) Acc. Chem. Res., 2004, 37(8), special issue on organocatalysis; (d) J. Seaved and B. List, Org. Biomol. Chem., 2005, 3, 719; (e) B. List and J.-W. Yang, Science, 2006, 313, 1584; (f) B. List, Chem. Commun., 2006, 819; (g) M. J. Gaunt, C. C. C. Johansson, A. McNally and N. C. Vo, Drug Discovery Today, 2007, 2, 8; (h) P. I. Dalko, Enantioselective Organocatalysis, Wiley-VCH, Weinheim, (i) Chem. Rev., 2007, 107(12), special issue on organocatalysis.
- 2. For recent reviews on organocatalytic cascade and related reactions see e.g.: (a) D. Enders, C. Grondal and M. R. M. Hüttl, Angew. Chem., Int. Ed., 2007, 46, 1570; (b) G. Guillena, D. J. Ramón and M. Yus, Tetrahedron: Asymmetry, 2007, 18, 693; (c) D. J. Ramón and M. Yus, Angew. Chem., Int. Ed., 2005, 44, 1602. See also e.g.: (d) D. Enders, M. R. M. Hüttl, C. Grondal and G. Raabe, Nature, 2006, 441, 861(e) D. Enders, M. R. M. Hüttl, J. Runsink, C. Raabe and B. Wendt, Angew. Chem., Int. Ed., 2007, 46, 467(f) Y. Hayashi, T. Okano, S. Aratake and D. Hazelard, Angew. Chem., Int. Ed., 2007, 46, 4922(g) H. Y. Kim, A. E. Lurain, P. Garcia-Garcia, P. J. Carroll and P. J. Walsh, J. Am. Chem. Soc., 2005, 127, 13138(h) N. Halland, P. S. Aburel and K. A. Jørgensen, Angew. Chem., Int. Ed., 2004, 43, 1272.
- 3. E. Reyes, H. Jiang, A. Milelli, P. Elsner, R. G. Hazell and K. A. Jørgensen, Angew. Chem., Int. Ed., 2007, 46, 9202.
- 4. (a) R. Takagi, Y. Miwa, T. Nerio, Y. Inoue, S. Matsumura and K. Ohkata, *Org. Biomol. Chem.*, 2007, **5**, 286; (b) J. Qi and J. A. Porco, Jr, *J. Am. Chem. Soc.*, 2007, **129**, 12682.
- 5. (a) J. K. F. Geirsson, J. F. Johannesdottir and S. Jonsdottir, Synlett, 1993, 133; (b) J. K. F. Geirsson and J. F. Johannesdottir, J. Org. Chem., 1996, 61, 7320; (c) K. Aoyagi, H. Nakamura and Y. A. Yamamoto, J. Org. Chem., 1999, 64, 4148; (d) J. K. F. Geirsson and E. G. Eiriksdottir, Synlett, 2001, 664; (e) J. K. F. Geirsson, L. Arnadottir and S. Jonsson, Tetrahedron, 2004, 60, 9149.
- 6. J. K. F. Geirsson, S. Jonsson and J. Valgeirsson, Bioorg. Med. Chem., 2004, 12, 5563.
- 7. For design and introduction of this catalytic system see: e.g.: (a) M. Marigo, T. C. Wabnitz, D. Fielenbach and K. A. Jørgensen, *Angew. Chem., Int. Ed.*, 2005, **44**, 794. See also: (b) M. Marigo, D. Fielenbach, A. Braunton, A. Kjærsgaard and K. A. Jørgensen, Angew. Chem., Int. Ed., 2005, 44, 3703; (c) Y. Hayashi, H. Gotoh, T. Hayashi and M. Shoji, Angew. Chem., Int. Ed., 2005, 44, 4212; (d) J. Franzén, M. Marigo, D. Fielenbach, T. C. Wabnitz, A. Kjærsgaard and K. A. Jørgensen, J. Am. Chem. Soc., 2005, 127, 18296; (e) C. Palomo and A. Mielgo, Angew. Chem., Int. Ed., 2006, 45, 7876
- 8. Crystal data for [4i]†: $C_{23}H_{25}BrO_{10}$, M = 540.06, orthorhombic, space group $P2_12_12$ (no. 112), a=15.9364(7) Å, b=15.936(0) Å, c=18.6763(10) Å, V=4743.2(3) Å, T=293 K, T=21.516 g cm⁻³, μ (Mo K α , $\lambda = 0.71073$ Å) = 1.789 mm⁻¹, 16280 reflections collected, 5229 unique [$R_{int} = 0.0296$], which were used in all calculations. Refinement on F2, final R(F) = 0.044, wR(F2) = 0.0808.
- 9. Note added in proof: O. Penon, A. Carlone, A. Mazzanti, M. Locatelli, L. Sambri, G. Bartoli and P. Melchiorre, Chem.-Eur. J., 2008, 14, 4788.